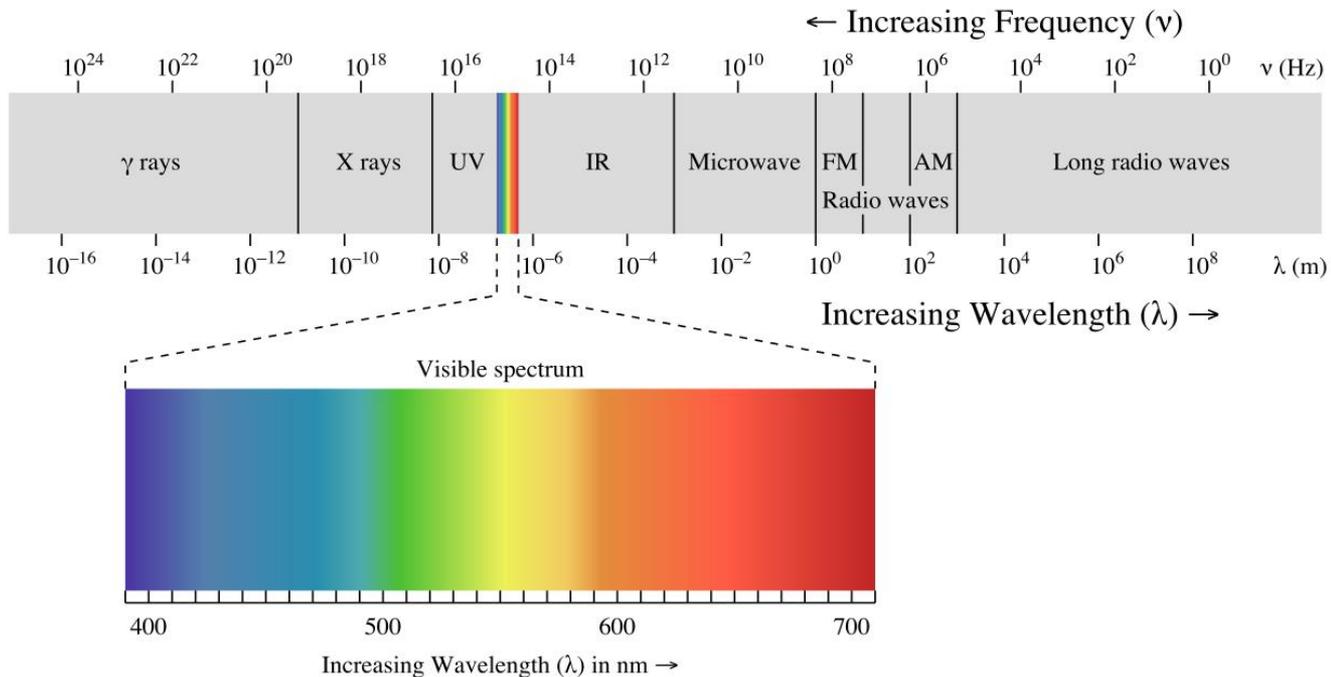


Optical Spectroscopy:

The study absorption and emission of light in nature

Abid Khan (past slides from Virginia Lorenz and Kai Wen Teng)

PHYS 403 Spring 2019

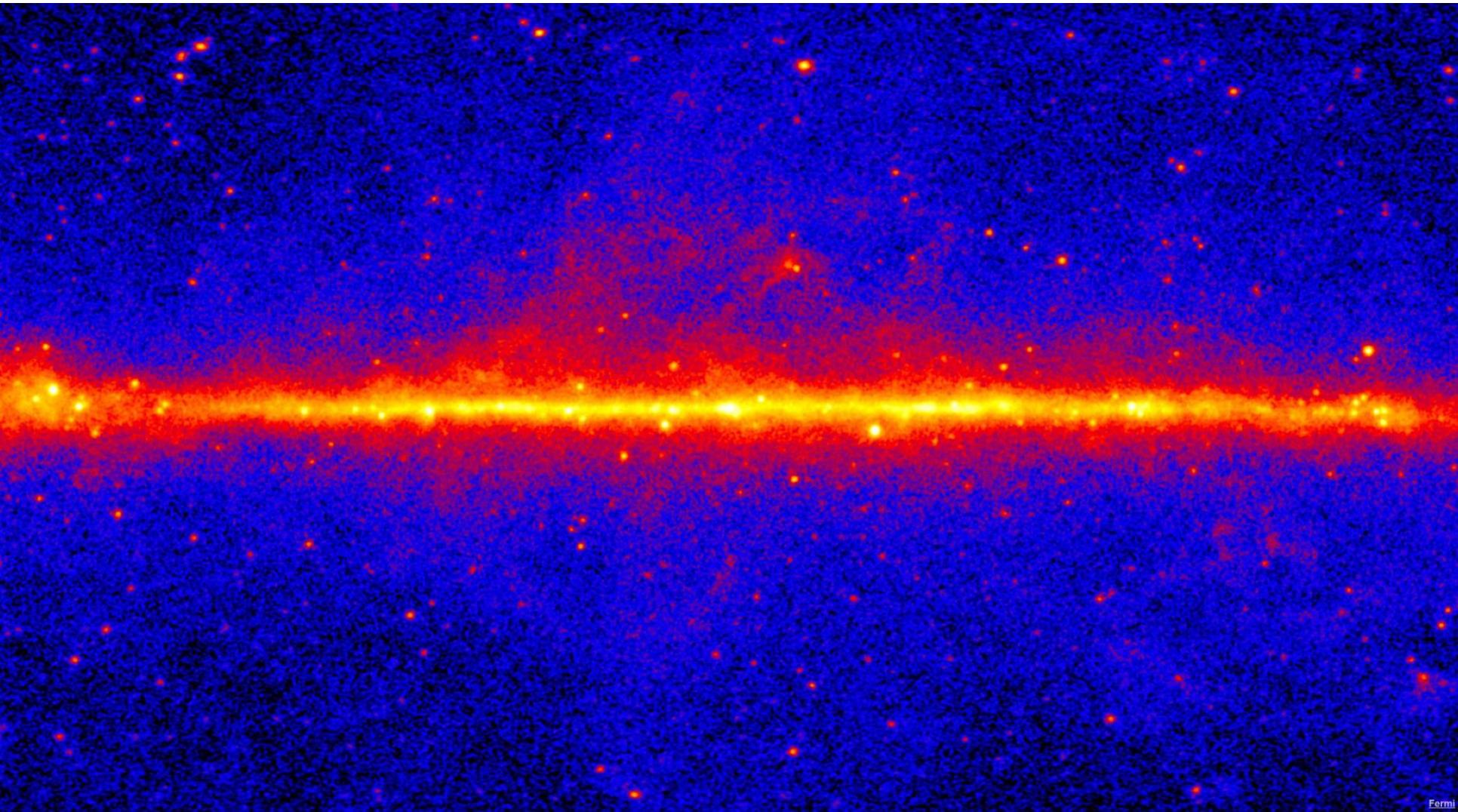


Optical Spectroscopy in Astronomy

- By looking through a standard telescope, you are observing the night sky at the visible light spectrum



Gamma Rays



Black hole binaries in disks of plasma are sources of gamma rays

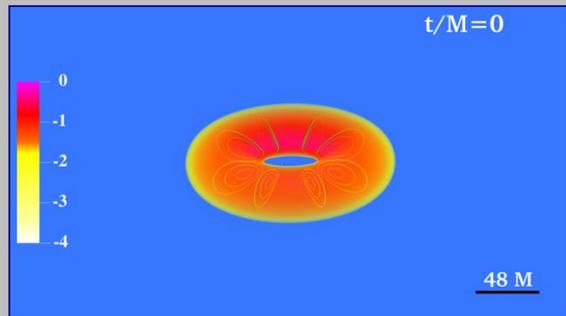


Fig. 2-1: *Initial Configuration*

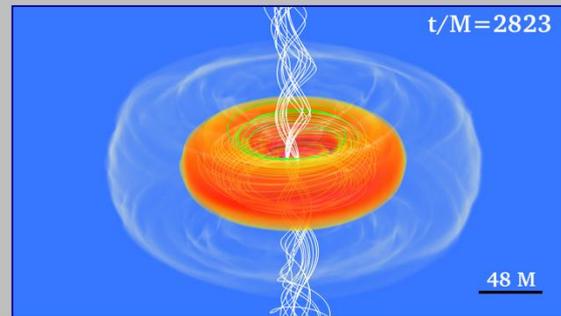


Fig. 2-2: *Twin jets rise from the inspiraling black holes*

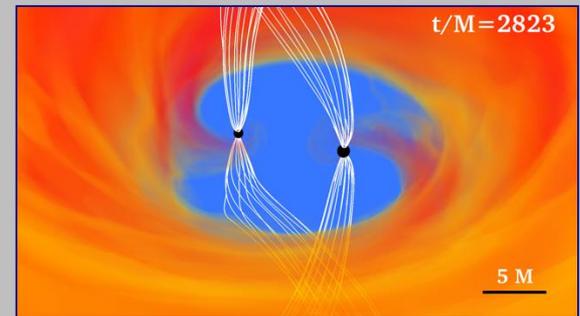


Fig. 2-3: *Zoom in to central cavity*

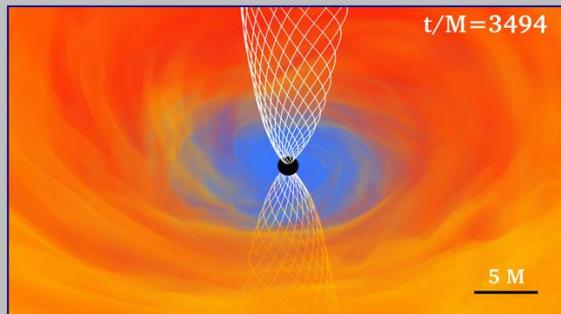


Fig. 2-4: *Post merger black hole and magnetic field*

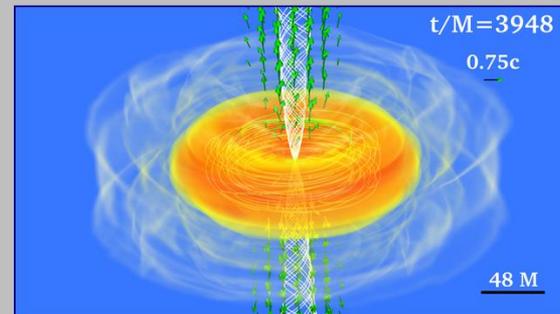
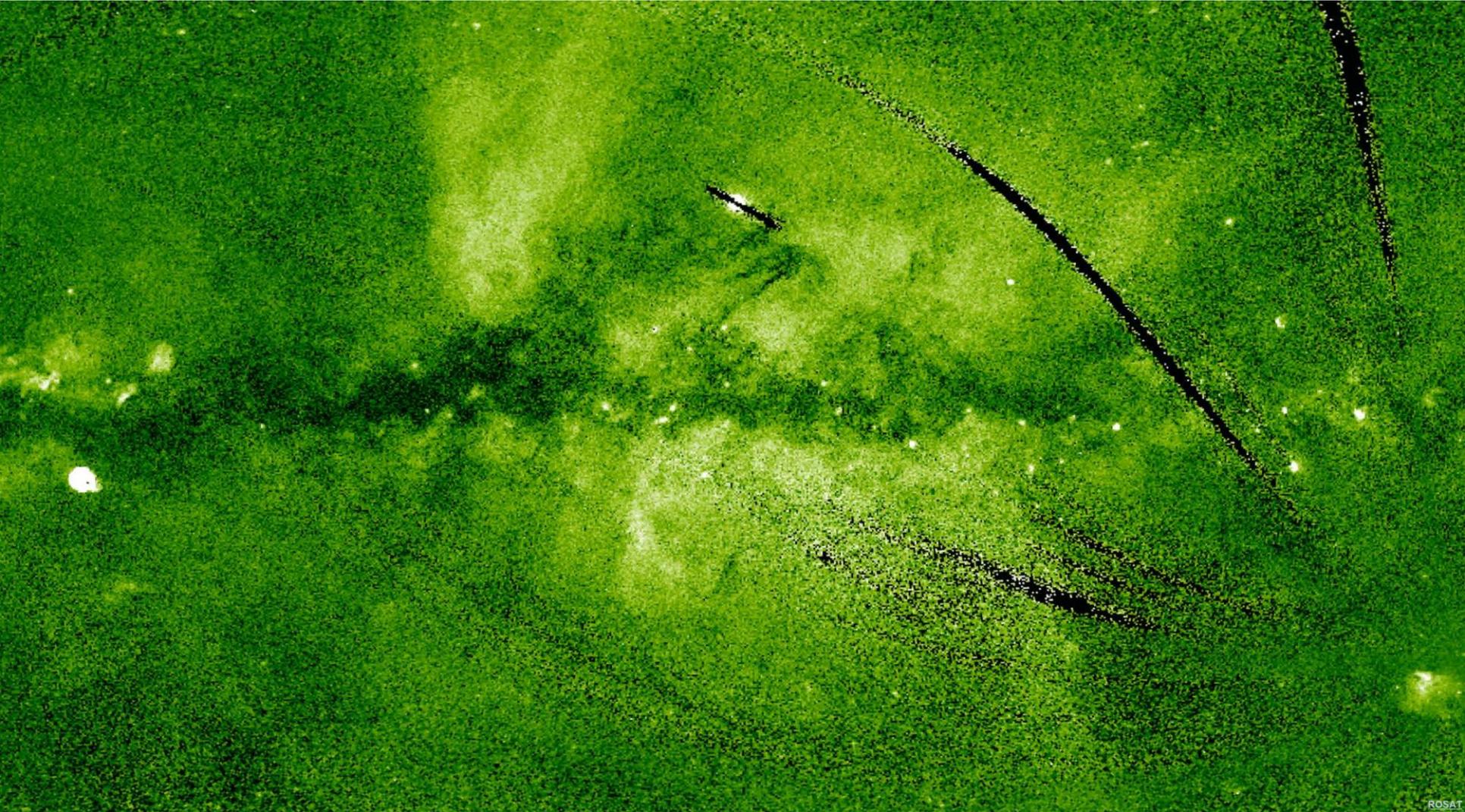


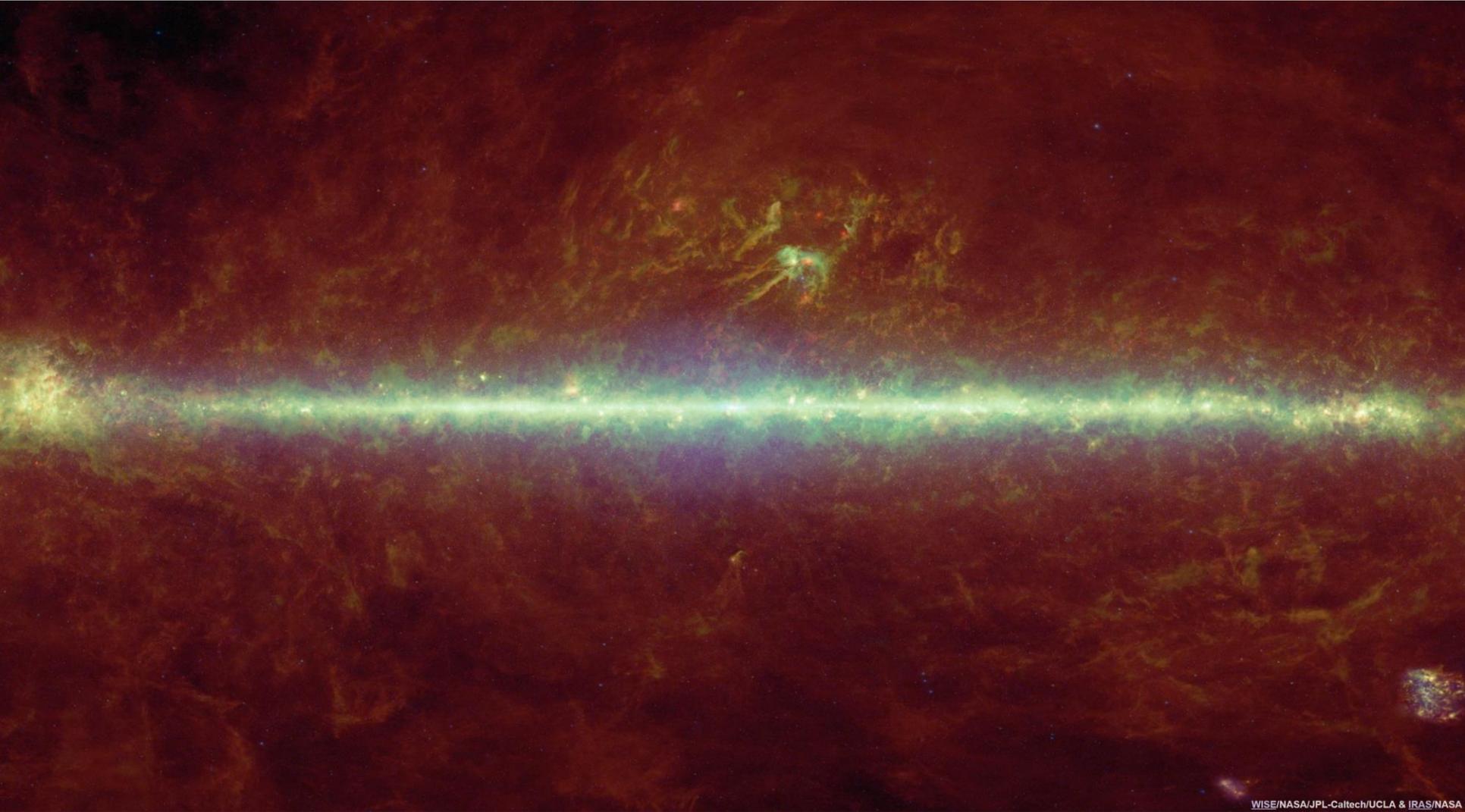
Fig. 2-5: *Quasistationary jet outflow*

X Rays



X-ray sources include stars, supernova, gaseous shells ejected during a violent explosion of a dying star, and synchrotron radiation

Infrared



WISE/NASA/JPL-Caltech/UCLA & IRAS/NASA

Sourced from stars

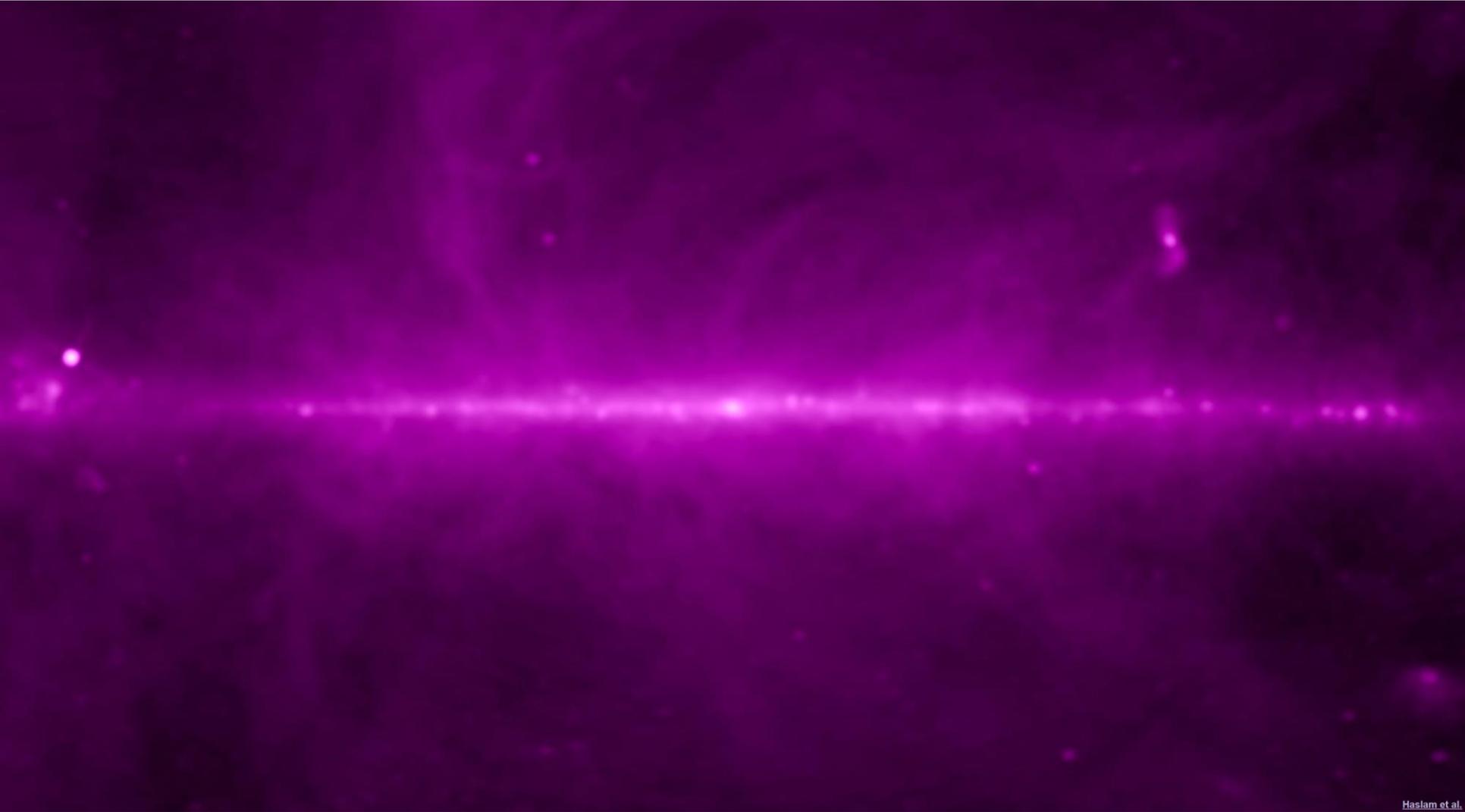
Microwave



ESA Planck LFI and HFI Consortia (2010)

Cosmic microwave background radiation emitted from the big bang and inflation

Radio



Haslam et al.

cold intergalactic hydrogen mostly found in the milky way emits radio waves.

Luminescence: Emission of light from any substance

- **Fluorescence:** transition from excited state to ground state is fast (~ns – ms range)
- **Phosphorescence:** transition from excited state to ground state is slow (~s – ks range)

Perrin-Jablonski energy diagram

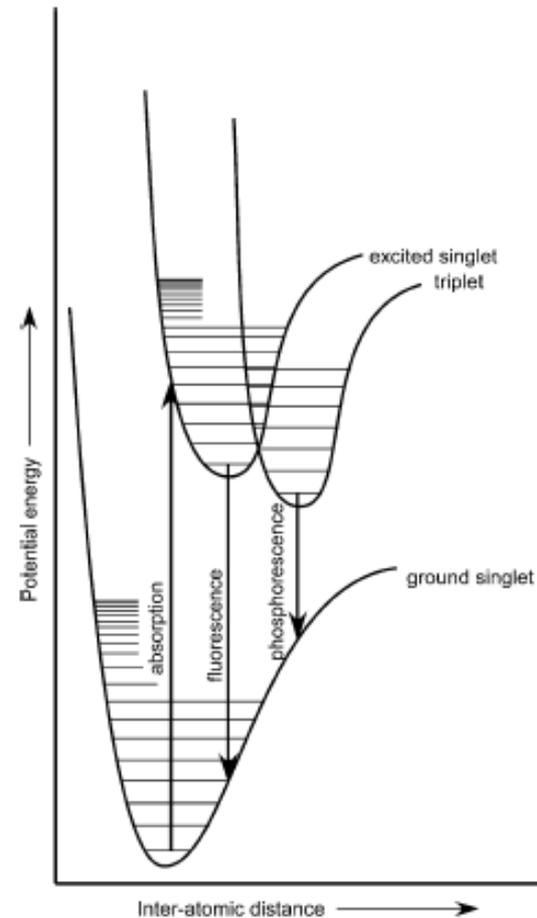
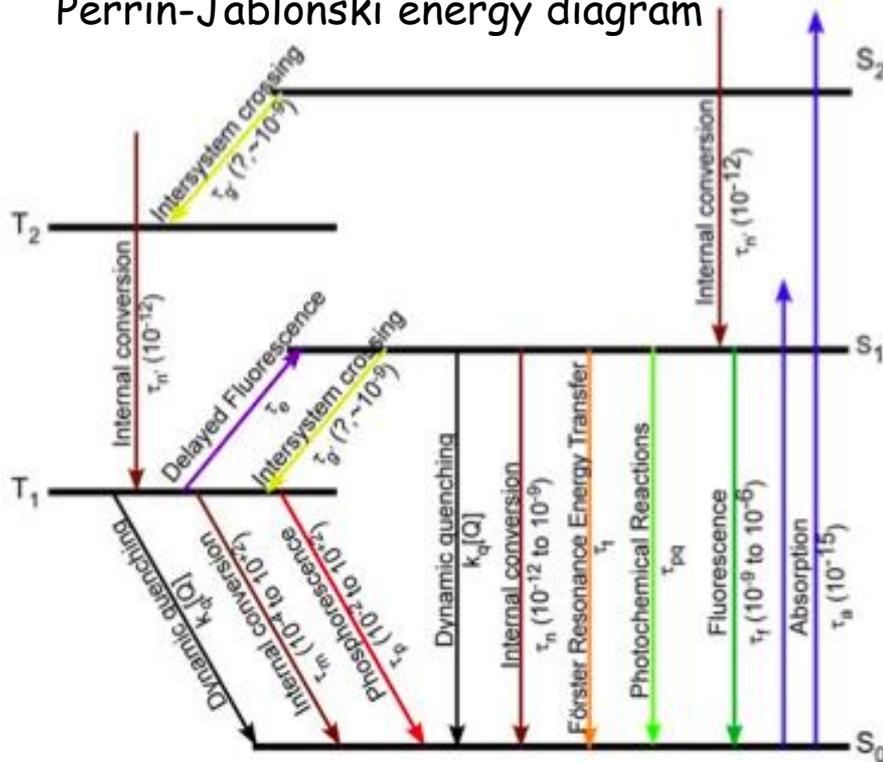
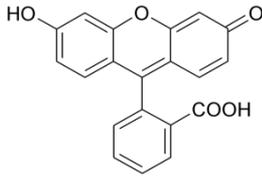


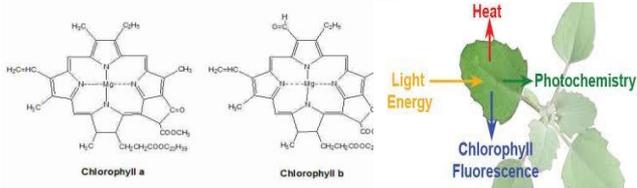
Diagram by Robert Clegg in Photosynth Res. 2009 Aug-Sep;101(2-3):181-94.

Types of Fluorescent Molecules

Synthetic Organic:
Fluorescein



Naturally Occuring:



Semiconductor Nanocrystal:

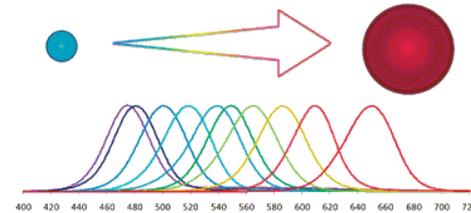
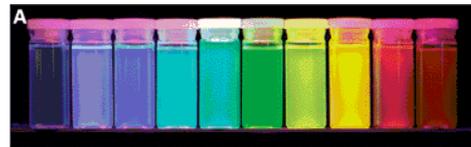
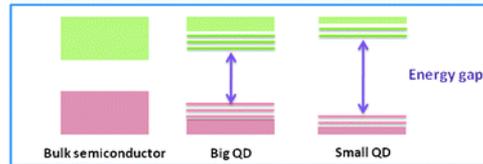
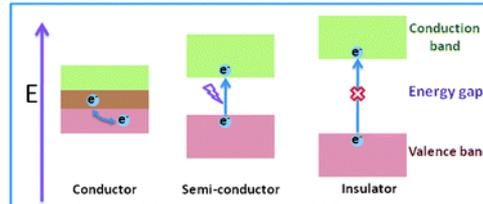
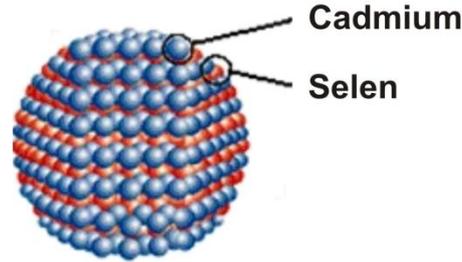
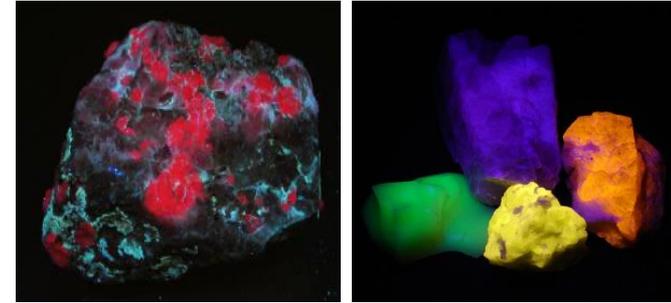


Image from Zrazhevskiy et al. 2010

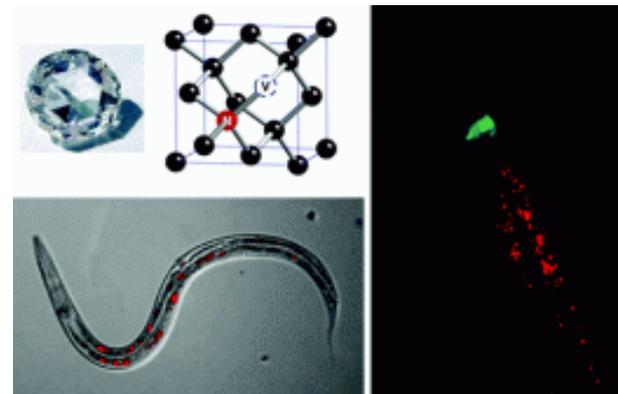
Crystals:



Ruby and assorted minerals

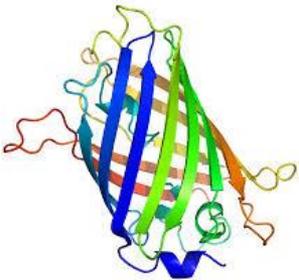
From mineralman.net

Fluorescent Nanodiamonds



Nano Lett., 2010, 10 (9), pp 3692-3699. DOI: 10.1021/nl1021909

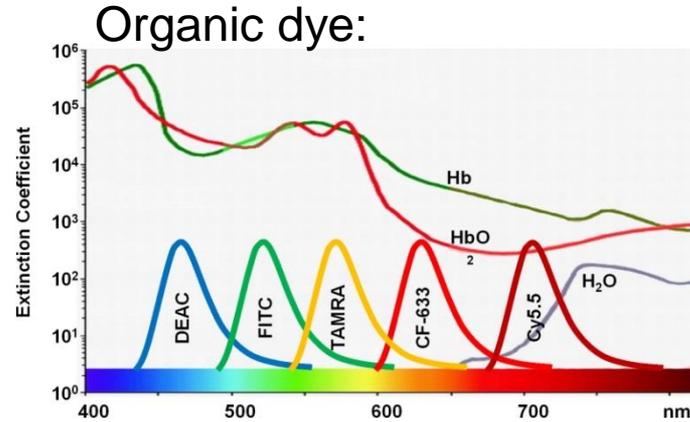
Fluorescent Proteins:



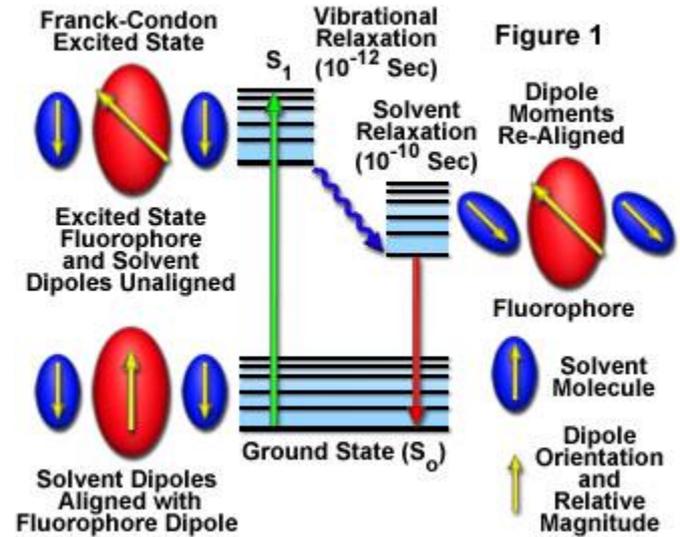
Green Fluorescent Protein

Fluorescence (S_1-S_0)

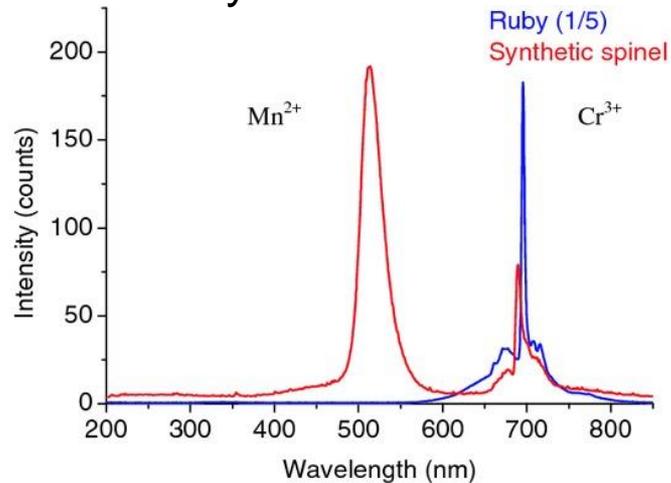
Solvent Effect:



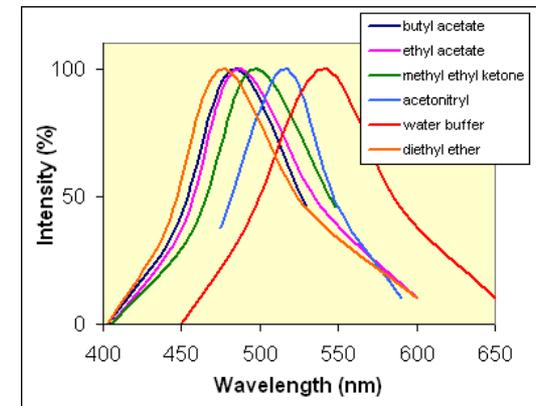
Fluorophore-Solvent Excited State Interactions



Ruby:



<http://micro.magnet.fsu.edu>

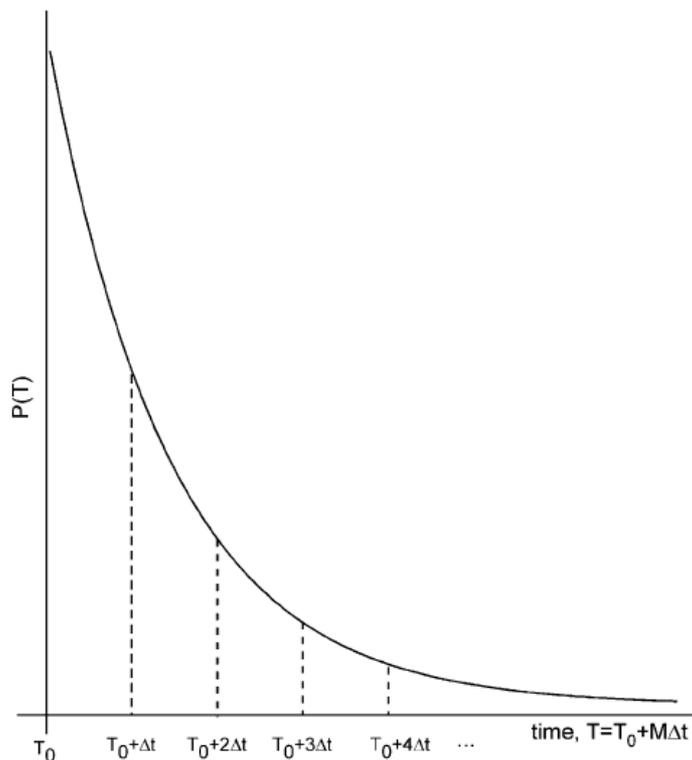


<http://www.bio.davidson.edu>

Time-Dependent Fluorescence: Fluorescence Lifetime

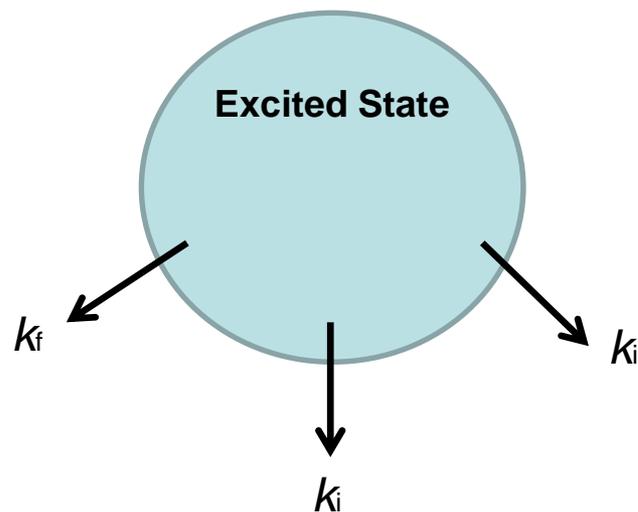
Fluorescence Lifetime: The average amount of time a molecule stays in excited state

Probability of being in the excited state



k_f = rate constant for leaving excited state while emitting a photon

k_i = rate constant for leaving excited state through other means (ie. Dynamic quenching, Energy Transfer, etc)



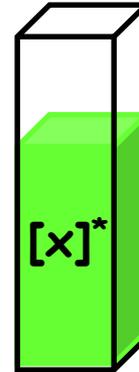
Fluorescence Lifetime:
$$\tau = \sum_i \frac{1}{k_i}$$

Lifetime is sensitive to other decaying pathways present!

Measuring the Depletion of the excited state

$$[\# x^*] = [\# x_o^*] e^{-(k_F + k_t)t}$$

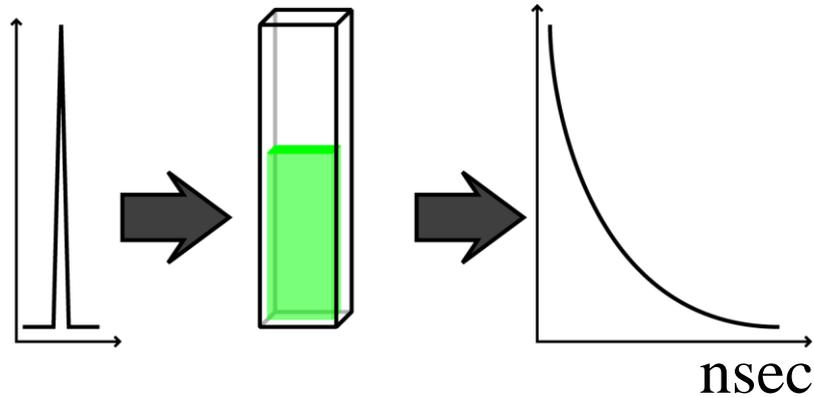
$$[\# x^*](k_F) = \text{Intensity that you measure}$$



k_F is rate constant of fluorescence

Intensity measured is proportional to the # of molecules in the excited state!

Measuring Lifetime: Time Domain

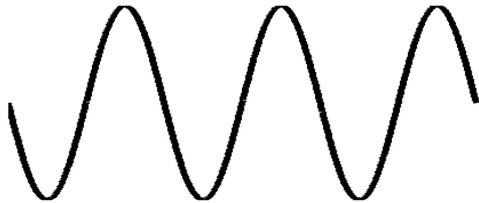


What do you need?

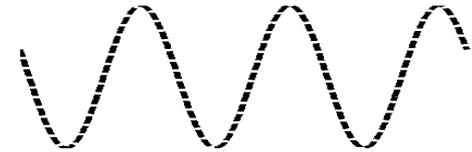
- Collect signal fast enough
- Fitting

Measuring Lifetime: Frequency Domain

$$E(t) = E_o + E_\omega \cos(\omega_E t + \varphi_E)$$



$$F(t) = F_o + F_\omega \cos(\omega_E t + \varphi_E - \varphi)$$



$$\tan(\varphi) = \omega_E \tau_\varphi$$

$$M = \frac{F_\omega / F_o}{E_\omega / E_o} = \frac{1}{\sqrt{1 + (\omega \tau_{Mod})^2}}$$

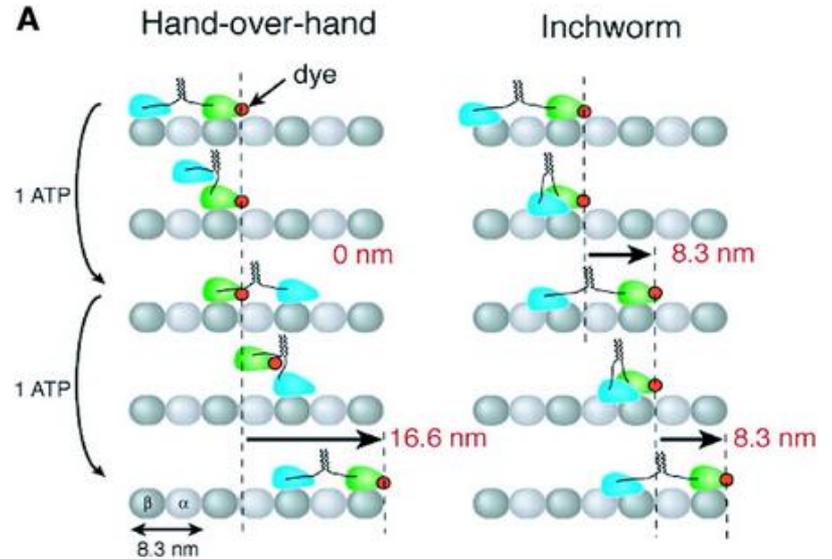
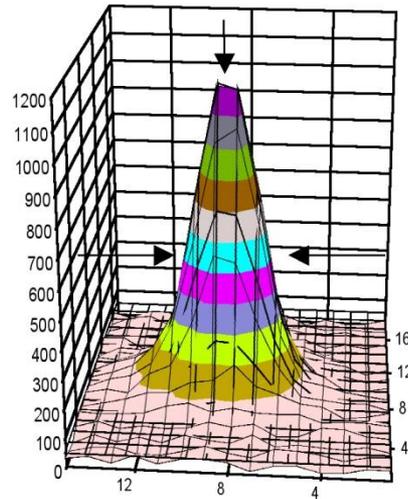
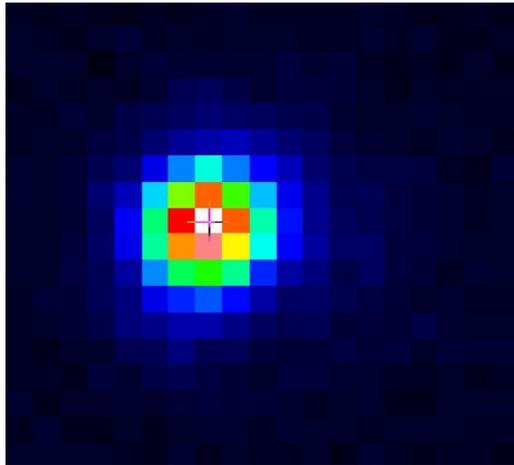
What do you need?

-Intensity modulators

-Synchronization

Applications of Fluorescence in Biology

Single Molecule Fluorescence Imaging (myosin)



$$\sigma_{\mu_i} = \sqrt{\left(\frac{s_i^2}{N} + \frac{a^2/12}{N} + \frac{8\pi s_i^4 b^2}{a^2 N^2}\right)}$$

Center of the distribution can be determined in ~1.5 nm accuracy if #N is more than 10⁴

